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### Characterization and Vibrational Properties of Undoped and AsF<sub>5</sub>-Doped (C<sub>6</sub>H<sub>4</sub>)<sub>x</sub>

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CHARACTERIZATION AND VIBRATIONAL PROPERTIES OF UNDOPED AND  
AsF<sub>5</sub>-DOPED (C<sub>6</sub>H<sub>4</sub>)<sub>x</sub>

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**Abstract** A vibrational study of poly-p-phenylene (PPP) prepared by the Kovacic method is presented. A comparison is made between the vibrational features (in both IR and Raman) of undoped PPP, PPP chemically doped using AsF<sub>5</sub> or electrochemically doped with AsF<sub>6</sub><sup>-</sup>, and finally dedoped by a thermal treatment. In the last case, using the exciting wavelength  $\lambda_L = 647.1$  nm, additional modes are observed in the Raman spectra (460 cm<sup>-1</sup>, 920 cm<sup>-1</sup> and 1700 cm<sup>-1</sup>). In AsF<sub>5</sub> doped PPP, the IR absorption spectrum reveals also new modes (1552 cm<sup>-1</sup>, 1273 cm<sup>-1</sup> and 1184 cm<sup>-1</sup>). These features are discussed in the basis of a vibrational model.

### I UNDOPED PPP (Kovacic)

Raman spectra of undoped PPP are always perturbed by a very strong fluorescence with a maximum peaked around 600 nm, and which depends upon the synthesis of the material as well as upon the washing procedure. Catalyst residues, for example, lead to a slight doping of the PPP powder, and quinoid structures or other defects are expected to be created. The catalyst residues contents can, according to ref. (1), give rise to polymer powders of different colors (dark brown or clear brown), whose infrared spectra are similar except a band at 1600 cm<sup>-1</sup>. It is reported that the dark brown powder may contain more quinoid structure in the chain, leading to a shift to 385 nm of the optical absorption band (2) while the maximum absorption of a p-polyphenyl sample is predicted

at 339 nm when  $n$  is infinite (3).

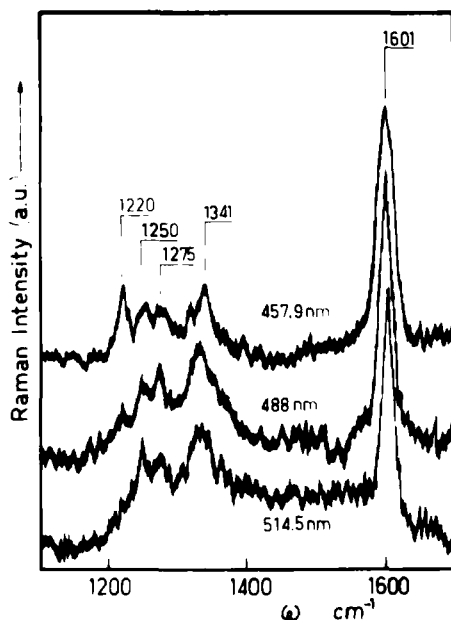


FIGURE 1 Raman spectra of PPP (Kovacic) at  $T = 10$  K for  $\lambda_L = 457.9$  nm, 488 nm and 514.5 nm.

In figure 1 are shown the Raman spectra of a PPP prepared by the Kovacic procedure (dark brown powder) for different excitation wavelengths, in which the fluorescence background has been subtracted. They are composed of a strong band at  $1600\text{ cm}^{-1}$  and many weaker bands at 1220, 1250, 1275 and  $1341\text{ cm}^{-1}$ .

## II AsF<sub>5</sub>-DOPED PPP

Raman spectra of AsF<sub>5</sub>-doped PPP have been reported previously (4). Upon doping, the fluorescence disappears completely and three Raman bands are mainly observed :  $1598\text{ cm}^{-1}$ ,  $1320\text{ cm}^{-1}$  and  $1236\text{ cm}^{-1}$  for  $\lambda_L = 647.1$  nm. The mode at  $1598\text{ cm}^{-1}$  does not depend on  $\lambda_L$ . On the contrary, going to  $\lambda_L = 647.1$  nm to  $\lambda_L = 406.7$  nm, it must be no-

ticed that the  $1236$  and  $1320\text{ cm}^{-1}$  decrease in intensity while two other modes at  $1216$  and  $1275\text{ cm}^{-1}$  appear, resulting in an apparent shift. By electrochemical doping, similar results are roughly observed, the Raman spectra of PPP doped electrochemically being composed of modes at  $1224$ ,  $1280$  and  $1595\text{ cm}^{-1}$  respectively.

### III THERMAL STABILITY OF $\text{AsF}_5$ -DOPED PPP

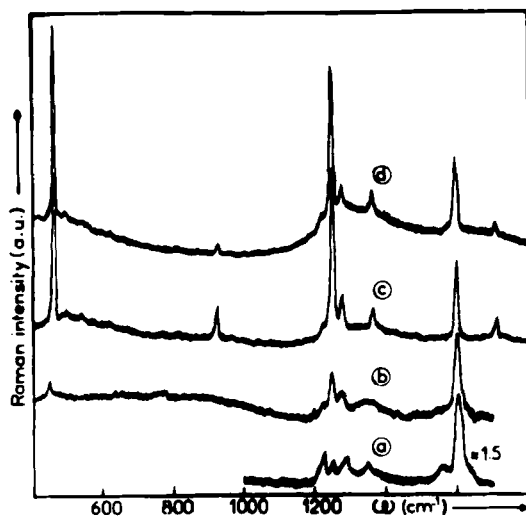


FIGURE 2 Raman spectra at 78 K of PPP Kovacic - doped chemically with  $\text{AsF}_5$  and dedoped by heating at  $250^\circ\text{C}$  during 15 mn.

a)  $\lambda_L = 457.9\text{ nm}$  ; b)  $\lambda_L = 514.5\text{ nm}$   
 c)  $\lambda_L = 647.1\text{ nm}$  ; d)  $\lambda_L = 676.4\text{ nm}$

An  $\text{AsF}_5$  doped PPP sample has been heated at several temperatures from  $100^\circ\text{C}$  to  $250^\circ\text{C}$ . Until  $200^\circ\text{C}$ , the sample does not show any apparent modification but its color goes from green to blue at  $250^\circ\text{C}$ . During this treatment, new modes appear at  $920$  and  $1700\text{ cm}^{-1}$  as shown on Figure 2. In addition, a mode at  $460\text{ cm}^{-1}$ , which was very weak in the doped sample, is now very strong in intensity. Also, the mode at  $1240\text{ cm}^{-1}$  becomes very strong. All

these new features are observed for  $\lambda_L = 647.1$  nm or 676.4 nm. When we use a higher energy excitation ( $\lambda_L = 457.9$  ou 514.5 nm), the Raman spectrum is very much like that of the undoped PPP (Figure 2).

#### IV CONCLUSION

The problem whether a quinoïd structure exists or not in the PPP chain will certainly be resolved from the vibrational studies. Concerning the Raman spectroscopy, a very strong resonant behavior is detected with new strong modes appearing in the red range excitation. This resonance is connected with the absorption in this region and the differences existing in samples with and without contact with air and moisture (5). Moreover, we showed previously (4) that, in the Raman spectra of oligomers and PPP prepared by the Yamamoto method, the Raman modes at 1220 and 1275  $\text{cm}^{-1}$  do not depend on  $\lambda_L$ , while the 1246 and 1350  $\text{cm}^{-1}$  modes are not seen. Therefore, the latter Raman modes could be connected to a quinoïd structure in PPP chains prepared by the Kovacic method. It must be noticed that the assignment of the different modes is under way with a vibrational model, as well as with that predicted recently by Zannoni and Zerbi (6).

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